Electrochemistry Half-Cell Plate

Order Code ECHEM-PLT

The Electrochemistry Half-Cell Plate is a reusable, premeasured platform designed to simplify electrochemical experiments with multiple metal-ion half cells. Its innovative design allows for precise measurements and easy setup, making it an efficient tool for studying electron transfer reactions and electrochemical processes.



Note: Vernier products are for educational use only.

Using the Product

Before Each Use

- 1. **Fill the Wells**: Add 1.0 M KNO₃ solution to both the central salt bridge well and each half-cell well.
- 2. **Soak Overnight**: Let the module soak overnight to ensure the salt bridge and porous cylinders are fully saturated for optimal ion transfer.
- 3. **Empty the Wells**: Before starting the experiment, empty the KNO₃ solution from all wells to prepare them for fresh solutions.

During Use

- Refill the Salt Bridge Well: Fill the central salt bridge well with fresh 1.0 M KNO₃ solution.
- 2. **Prepare Electrolyte Solutions**: Select and prepare the appropriate electrolyte solutions for your experiment.
- 3. **Fill the Half-Cell Wells**: Fill each half-cell well with the selected electrolyte solutions, ensuring they are filled to the overflow rim. Make sure the lower channel leading to the salt bridge is fully saturated.
- 4. **Insert Metal Electrodes**: Place the polished metal electrodes into the corresponding half-cell wells.
- 5. **Measure Voltage**: Use the Go Direct® Voltage Probe or another measurement device to record the voltage between the half-cells and analyze the electrochemical reactions. An example of a final experimental set-up is shown in Figure 1.

Figure 1

After Use

- 1. **Remove Metals**: Take out the metal electrodes.
- 2. **Dispose of Liquids**: Properly dispose of all used liquids according to lab protocols.
- Rinse the Module: Rinse all wells and components with deionized water to clean them before storage.

Care and Maintenance

Ensure you only use KNO₃ to soak the salt bridge in the central ring. Using other chemicals, especially NaCl, can cause irreversible reactions with the salt bridge, potentially rendering it unusable.

Storage

- If reusing soon, store the module in a humid environment (such as a sealed plastic bag) to prevent the need to recharge the salt bridge with ion solution.
- If not needing to use again soon, store the module dry.
- Be sure to thoroughly rinse all wells with deionized water before storage. Any remaining metal ions can compromise the performance of future experiments.

Accessories/Replacements

 Item
 Order Code

 Electrochemistry Metals Kit
 ECHEM-MTLS

 Go Direct Voltage
 GDX-VOLT

Warranty

Warranty information for this product can be found on the Support tab at http://www.vernier.com/echem-plt

General warranty information can be found at www.vernier.com/warranty



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